Chelating ion exchange resins selective for boron compounds. (Unitika Ltd., Japan). Jpn. Kokai Tokkyo Koho (1983), 6 pp. CODEN: JKXXAF JP 58146448 A2 19830901 Showa. Patent written in Japanese. Application: JP 82-28655 19820223. CAN 100:8112 AN 1984:8112 CAPLUS

# **Patent Family Information**

ind Date	Application No.	Date
2 19830901	JP 1982-28655	19820223
4 19880114		
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19820223		•
	2 19830901	2 19830901 JP 1982-28655 4 19880114

## **Abstract**

The title resins are 3-dimensional network polymers of phenols, aldehydes, amino polyols, and aliph. polyamines. Thus, PhOH 29.1, 37% aq. HCHO 25.0, and N-methyl-D-glucamine 60.0 g were combined in water and heated to 80° for 2 h, after which 22% aq. NaOH 56.1, ethylenediamine 11.2, and 37% aq. HCHO 100 g were added and the mixt. stirred 1 h at 30°, followed by addn. of 33.9 g resorcin and 113 g 37% aq. HCHO while cooling the mixt. to 5-20°, then dispersion in C2Cl4, pearl polymn., and workup to obtain 180 g dark brown resin beads contg. 50% absorbed water. When 1 mL of the wet resin beads (titrated with HCl to phenolphthalein endpoint) were added to 50 mL of concd. brine contg. 11.8 mg/L B and shaken for 24 h at 25°, the B content of the brine fell to 0.3 mg/L.





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# JP58146448A2: CHELATING ION EXCHANGE RESIN AND ITS MANUFACTURE AND ADSORPTION **REATMENT** ØTitle:

Chelating ion exchange resin - comprises three dimensional crosslinked condensn. polymer of 

phenoll, aldehyde amino:poly:ol and aliphatic poly:amine [Derwent Record]

JP Japan P Country:

A (See also: JP63001897B4) ® Kind

WAYA YOSHIAKI; PInventor:

**UNITIKA LTD** 

News, Profiles, Stocks and More about this company PAssignee:

**1983-09-01** / 1982-02-23 Published / Filed:

JP1982000028655 P Application

Number

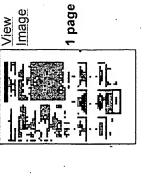
IPC-7: B01J 45/00; B01J 47/00; C08G 14/06; PIPC Code:

1982-02-23 JP1982000028655 Priority Number: PURPOSE: To obtain an entitled resin having superior selective & Abstract:

adsorption speed, and superior mechianical strength and resistance to org. staining, by resinifying a compd. obtd. by introducing an adsorptivity especially for boron, hydrophilic property, high

aminopolyalcohol deriv. to a phenol ring.

adduct. As this aminoalcohol, compds. represented by the formula aldehydes, such as formaldehyde, and one of aminopolyalcohols are reacted to obtain an early product of phenol-aminopolyalchol glucamine or n-ethyl-d-glucamine are used. Aldehydes, aliphatic CONSTITUTION: One of phenols, such as phenol, one of in which n is an integer of 1W10, embodied as N-methyl-D.



polyamines, and phenols are added to said early stage product, and they are polycondensed in the presence of an alkaline catalyst to obtain an intended chelating ion exchange resin having a 3-dimensionally cross-linked structure.

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